Effect of urea-N on growth and indoleacetic acid production of Stenotrophomonas maltophilia (Sb16) isolated from rice growing soils in Malaysia

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INTRODUCTION

Rice (Oryza sativa L.) is one of the major staple food crops in the world (Wang et al., 2004), and N is the most important input for high yield. It is known that rice requires 1 kg N to produce 15-20 kg of grain. Yields per hectare are critically dependent on the nature, and amount and timing of N supply (George et al., 1992). Urea is the most commonly used N fertilizer in rice cultivation, and timing of N supply (George et al., 1992). Urea is produced in granular or pellet forms, and is coated with a non-hygroscopic inert material. After applied to the soil, the highest commonly used N fertilizer in rice cultivation, produced in granular or pellet forms, and is coated with a non-hygroscopic inert material. After applied to the soil, its N is rapidly changed into ammonia.

Wetland rice ecosystem is unique where N can be supplied to plant through biological N fixation (BNF). Rice plant naturally associates with free living N supplied to plant through biological N fixation (BNF). Wetland rice ecosystem is unique where N can be supplied to plant through biological N fixation (BNF). Rice plant naturally associates with free living N supplied to plant through biological N fixation (BNF).

Excessive use of N fertilizers is now known to affect the ecosystem, bacterial N fertilizer production. The importance of BNF technology can play a role in substituting commercially available N fertilizer use in rice culture. The rates of N2 fixation by free-living and associative diazotrophs in rice fields are low compared with the rates of N2 fixation by legumes (Bolholm et al., 1992). However, recent study shows that free living diazotrophs can supplement about 40% of total N to the rice (Naher et al., 2011). Previous study reported that high levels of mineral N caused a significant decrease in the N2 fixation as measured by acetylene reduction activity (ARA) due to the inhibition of nitrogenase enzyme synthesis (Junior et al., 2000). The process of N2 fixation by free-living bacteria as well as by symbiotic associations is inhibited in the presence of fixed N especially ammonium.

Association of diazotrophs improves plant growth by several mechanisms and one of the mechanisms is production of phytohormone such as indoleacetic acid (IAA). Application of N fertilizer is essential for high rice production but excessive use of N may bring adverse effects on the ecosystem, bacterial N2-fixing activity and IAA production. The objective of this study was to determine the effect of different levels of urea fertilizer on the growth, N2 fixation and indoleacetic acid production of local diazotrophic strain Sb16 in paddy soil.

MATERIALS AND METHODS

Inoculum preparation

The bacterial strain Stenotrophomonas maltophilia Sb16 (accession number JQ820255) was previously isolated from Tanjung Karang Rice growing area in Malaysia (Naher et al., 2008). The bacterial strain (Sb16) used was Gram negative rod, with cellulolytic enzyme activity, high IAA (60 mg L^-1), nitrogenase activity of 1.4 × 10^-7 µmol C2H2 l^-1 cfu^-1 h^-1 and 43% Nfda (Naher et al., 2009; 2011). Starter...
culture was prepared by growing pure culture of Sb16 in Jensen’s N-free broth for 36 h. Composition of broth (L⁻¹): 20.0 g sucrose, 1.0 g K₂HPO₄, 0.5 g MgSO₄·7H₂O, 0.5 g NaCl, 0.1 g FeSO₄, 0.005 g Na₅MoO₄, 2.0 g CaCO₃, pH was adjusted to 6.8-7.0. One milliliter of the starter culture containing approximately 2 × 10⁷ cfu mL⁻¹ was transferred to a 100 mL flask containing 50 mL of Jensen’s N-free broth medium and allowed to grow to exponential growth phase.

**Soil inoculation and incubation**

Bacteria cultures were harvested, washed with phosphate buffer solution (0.85% PBS), and immediately suspended into PBS solution. Before applied inoculum optical density (OD₆₀₀) of washed cells were checked and adjusted to 0.1 and the population was confirmed by cell enumeration using drop plate method on N-free media (Somasegaran and Hoben, 1985). Approximately 1 × 10⁹ cfu mL⁻¹ of live washed cells was applied in to each treatment using sterilized pipette. The soil used in the study contained 0.1% N and pH 6.3. About 100 g of autoclaved sieved (4 mm) soil was placed in 250 mL conical flask and was flooded with 100 mL sterilized distilled water to maintain a standing water of about 3 cm. The bacterial treatments contained five levels of urea-N (0, 50, 100, 150, and 200 kg ha⁻¹). The experimental units were kept in incubator maintaining temperature 28 ± 2 °C for 6 wk. The flasks were destructively sampled weekly. Bacterial population in soil and soil-standing water, total N and IAA production were determined.

**Determination of bacterial population**

For soil population, 10 g of soil from the bottom of the flask was transferred into a conical flask containing 95 mL sterilized distilled water, while, for soil-standing water, 1.0 mL water was transferred into flask containing 99 mL sterilized water. The mixture was shaken vigorously on a rotary shaker for 10 min to suspend bacterial cells. A serial dilution was prepared and population was determined using drop plate method in N-free medium (Somasegaran and Hoben, 1985). Composition of the medium is (L⁻¹): 5 g malic acid, 0.5 g K₂HPO₄, 0.2 g MgSO₄·7H₂O, 0.1 g NaCl, 0.02 g CaCl₂ and 0.5% bromothymol blue in 0.2 N KOH (2 mL), 1.64% Fe-EDTA solution (4 mL) and 20 g agar. The change of media color from dark green to blue and growth of colony was indicating the N₂-fixing activity of the bacteria.

**Determination of soil/water pH and total N**

Soil pH and soil-standing water were determined at weekly intervals. For total N determination soil and soil-standing water samples were digested (H₂SO₄·H₂O₂ acid digestion) and total N was determined using Auto-analyzer (QuickChem 8000 series FIA System, Lachat Instrument, Loveland, Colorado, USA).

**Extraction of IAA from soil and water**

The IAA concentration of soil sample was determined using modified method of Sarwar et al. (1992). Three grams of soil were placed into a 50 mL Erlenmeyer flask and treated with 6 mL of phosphate buffer (0.2 M, pH 7.0) and 4 mL of L-tryptophan solution (5.3 g L-tryptophan kg⁻¹ soil). The flask was covered with parafilm and incubated in darkness at room temperature (± 28 °C) for 12 h on a shaker (~ 150 rpm). After incubation, flask contents were treated with 2 mL trichloroacetic acid (5 g 100 mL⁻¹ H₂O₂) to terminate the reaction and 1 mL calcium chloride (0.5 M) to facilitate filtration. The soil standing water was filtered through Whatman Filter paper nr 2. A buffer solution without incubation of soil was also prepared as a standard solution. For soil and soil standing water sample, approximately 2 mL water supernatant were mixed with Salkowski reagent (Gordon and Weber, 1951) and the mixture was allowed to stand 30 min for color development. The intensity of the color development was measured at 535 nm by using a spectrophotometer (Milton Roy, Rochester, New York, USA). The amount of L-tryptophan-derived auxins content in soil and soil standing water was determined as IAA-equivalents (mg kg⁻¹ soil) using standard IAA solution.

**Statistical analysis**

The experiment was conducted in factorial completely randomized design with three replicates. All experimental data were statistically analyzed by ANOVA using SAS (9.1 version) statistical software. Treatment means were compared using Tukey’s test (p ≤ 0.05).

**RESULTS AND DISCUSSION**

**Population of diazotrophs Sb16**

Application of different levels of urea-N increased population of Sb16. However, population differed significantly with time (Figure 1). In general bacterial population in the soil-standing water was higher than in soil. Significantly high population growth was observed in soil treated with 200 kg N ha⁻¹. Soil applied with urea-N showed higher population growth at first week and then decreased with increasing time. On the other hand, population in soil standing water was found high at second week of incubation. It is known that population of the diazotrophs can be affected by several factors including pH, temperature, nutrients, water, oxygen, and metabolic compounds (Döbereiner and Pedrosa, 1987). This study showed that population of Sb16 was indeed affected by the use of urea-N in the soil. The bacterial population decreased with increasing incubation time probably due to urea-N reduction over time and it was an important nutrient source for bacteria growth. Compared to soil, soil standing water maintained higher population as the applied bacteria contributed some of NH₄-N to the water. Previous study also showed that application of Sb16 increased NH₄-N level in soil water (Othman et al., 2012).

There was a positive significant relationship found between N rates and population of bacteria (Figure 2).
There were significant differences in population at first, third, and sixth week with increasing rates of urea-N. This could be due to the utilization of available N for cell growth. During the high bacterial growth in urea-N treatments, gelatinous material was observed to form in the soil layer of the incubation flask. The material could be the extra cellular polysaccharide produced by bacterial cells. Polysaccharide is a polymer that plays an essential role for bacterial growth and survival (Castro et al., 2008), it protects cell from desiccation and help in N\textsubscript{2} fixation by preventing high oxygen (O\textsubscript{2}) tension (Kumari et al., 2009).

### Soil chemical properties

The pH was significantly affected by different levels of urea-N (Figure 3). Soil pH increased after first week of incubation and stabilized thereafter. After the first

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**Figure 1.** Effect of urea-N on population of Sb16 in soil (a) and soil standing water (b). Bars indicate standard error n = 5.

**Figure 2.** Bacterial population at different levels of urea-N in soil at first week (a), third week (b), sixth week (c), and in soil standing water at first week (d), third week (e), and sixth week (f).
week of incubation, soil pH increased from 6.7 to 7 and pH of control soil was lower than the other treatments. Soil treated with urea-N had higher pH than control as presence of high NH$_4^+$ concentration in the soil increased pH. Optimal pH for N$_2$ fixation is 5-8 (Leigh, 2002). The consistencies of soil pH provide a stable environment for the growth of Sb16 that increases the survival and activity of Sb16 in soil and water.

The pH of soil-standing water was higher compared to pH of soil which ranged from 7.4 to 8.4 (Figure 3b). The pH change was probably due to the formation of ammonium ion in the soil standing water. Ammonium ion can also be formed through N$_2$ fixation by Sb16. The pH of soil and soil water in the first, third, and sixth weeks of incubation increased significantly with increasing rate of urea-N (Figure 4).

Figure 3. Effect of urea-N on soil pH (a) and soil standing water pH (b). Bar indicates standard error n = 3.

Figure 4. pH changes at different levels of urea-N in soil at first week (a), third week (b), sixth week (c), and in soil water at first week (d), third week (e), and sixth week (f).
There were significant effects of different levels of urea-N on total soil N content. Total N in soil decreased with increasing time of incubation (Figure 5). Total N content in soil and soil water decreased with increasing incubation time as it was used up by the applied bacteria. Other important factors may regulate NH\textsubscript{3} loss as the pH of the soil solution was high. However, an increasing trend of total N in the soil standing water observed at the 5\textsuperscript{th} and 6\textsuperscript{th} week of incubation which showed the biological N fixation activity.

**Concentration of IAA**

The IAA production by bacteria was significantly affected by urea-N levels. Significantly high amount of IAA was produced in the control treatment and lowest amount produced in the 200 kg ha\textsuperscript{-1} N applied treatment. The IAA concentration in the soil standing water was higher than in soil fraction. The initial IAA concentration in soil ranged from 1.5 to 2 mg g\textsuperscript{-1} and it was observed to decrease with increasing time of incubation (Figure 6). The amount of IAA produced in this study was comparatively low. Previously the Sb16 was shown to produce high amount of IAA in the presence of tryptophan (Naher et al., 2009). The low concentration of IAA could be due to the presence of low amount of precursor, L-tryptophan in the soil and water. The presence of IAA stimulating amino acid such as L-alanine, L-asparagine, and L-lysine have been reported to be present in root exudates, which stimulate formation of IAA in soils (Naher et al., 2008).

A significant decrease in IAA concentration in the soil and soil standing water found with increasing rates of urea-N (Figure 7) which might be due to less activity of the added microbes.

**CONCLUSIONS**

Application of different rates of urea-N significantly increased population of diazotrophic strain Sb16. The total N decreased until sixth week either it was used by the bacteria as a nutrient or NH\textsubscript{3} lost during incubation time. Application of urea-N significantly reduced the IAA production. The higher IAA value in control N treatments proved that higher doses of nitrogen reduced bacterial activities but did not hamper its growth.

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Figure 7. Effect of urea-N on indoleacetetic acid (IAA) production of Sb16 in soil at a) first week, b) second week, c) third week, and in soil water at e) first week, d) second week, and f) third week.

LITERATURE CITED


